

THE INFLUENCE OF EXOCYCLIC DNA ADDUCTS IN BACTERIAL AND MAMMALIAN GENOME INSTABILITY

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Summary

Oxidative stress enhances lipid peroxidation (LPO) implicated in the promotion and progression of carcinogenesis. One of the major LPO products is *trans*-4-hydroxy-2-nonenal (HNE), may to react with guanosine and under peroxidizing conditions also with adenosine. Additionally the same effect may induce environmental carcinogens, e.g. vinyl chloride and its metabolite chloroacetaldehyde (CAA). These compounds CAA and HNE introduce promutagenic exocyclic etheno and propano adducts into DNA, among them 1,*N*²-propanodeoxyguanine (PdG), 1,*N*⁶-ethenoadenine (1,*N*⁶-εA), 3,*N*⁴-ethenocytosine (3,*N*⁴-εC), *N*²,3-ethenoguanine (*N*²,3-εG) and 1,*N*²-ethenoguanine (1,*N*²-εG). CAA-induced additionally DNA damage in regions which revealed secondary structure perturbations rich in mutation hot-spots also in bacterial and mammalian genome. These perturbations may inhibited DNA synthesis and induced mechanisms of DNA repair such as BER or NER. Base excision repair constitutes the primary defense against lesions that do not heavily distort the DNA structure. BER is responsible for the removal of a variety of lesions. These include spontaneous hydrolytic depurination of DNA, deamination of bases, products of reaction with hydroxyl radicals, and covalent DNA adducts formed by intracellular LPO and small reactive metabolites, such as methylating agents. Repair is initiated by the action of a damage-specific DNA *N*-glycosylase that is responsible for the recognition and removal of an altered base through cleavage of the *N*-glycosylic bond and action of AP-endonuclease. Nucleotide excision repair (NER) is the most versatile and flexible DNA repair pathway of living cells as it deals with a wide range of structurally unrelated DNA lesions. NER corrects a wide array of DNA lesions that distort the DNA double helix, interfere in base pairing and block DNA duplication and transcription. The most common examples of these lesions are the cyclobutane pyrimidine dimers (CPDs) and 6-4 photoproducts (6-4 PPs) induced by ultraviolet radiation (UV) and bases with large substitutes derived from chemicals such as polycyclic aromatic hydrocarbons or exocyclic adducts.

Key words: DNA adducts, chloroacetaldehyde (CAA), lipid peroxidation (LPO), *trans*-4-hydroxy-2-nonenal (HNE), reactive oxygen species (ROS), base excision repair (BER), nucleotide excision repair (NER)

FORMATION OF EXOCYCLIC DNA ADDUCTS

Cellular DNA is continuously exposed to a variety of agents that alter its structure. These agents are both endogenous and exogenous, and include normal cellular metabolism, cell injury, inflammation, ionizing radiation and chemical agents. Accumulating evidence indicates that water, oxygen and endogenous alkylation are the main contributors to overall DNA damage (1).

These agents bring a considerable threat to living cells. Although both prokaryotic and eukaryotic cells are equipped with diverse DNA repair systems (2), removal of DNA lesions in an error-free way sometimes is not efficient enough and damage escapes processing before replication. Unrepaired DNA damage leads to various biological consequences, such as mutations or cell death, and subsequently to carcinogenesis, aging, and degenerative diseases (3).

Exocyclic DNA adducts are produced by endogenous and exogenous agents. Of the exocyclic DNA adducts, etheno (ε) bases have been the most widely studied over the last 25 years, as they are formed

by many genotoxic carcinogens, e.g., vinyl chloride or chloroacetaldehyde (4) and are also produced endogenously in animals and man. This class of DNA lesions affects normal Watson-Crick base pairing in DNA and was shown to be mutagenic in *E.coli* and mammalian cells (4).

It has been estimated that chronic inflammation is involved in the development of about one-fourth of all cancers worldwide. Inflammatory response leads to recruitment of activated leukocytes, which release high quantities of reactive oxygen species (ROS) such as superoxide and hydrogen peroxide. Hydrogen peroxide can produce hydroxyl radicals in reaction with metal ions. Direct proof comes from the work of Dizdaroglu (5) who showed that exposure of human cells to activated leukocytes causes DNA base modifications typical of hydroxyl radical attack. ROS also interact with membrane lipids causing their fragmentation and production of reactive aldehydes, which are able to interact with nucleic acids and form exocyclic DNA adducts. Etheno bases were first described by Kochetkov (6), who identified

them as fluorescent analogues for biochemical studies and probes for nucleic acid structures although, among different exocyclic adducts only 1, N^6 – ethenoadenine possesses fluorescent properties. The renewed interest in exocyclic DNA lesions in the 1990s was due to the development of ultrasensitive detection methods notably for etheno- and propano-DNA adducts which made it possible to study the formation of exocyclic adducts in experimental animals and humans. In 1994, unequivocal identification of the malondialdehyde-derived deoxyguanosine (M_1 -dG) adduct was reported by Chaudhary (7) in human liver. The same adduct was later also found in human breast and leukocytes. In 1995, Swenberg and co-workers found background levels of etheno- and propano-adducts in DNA of various human and rodent tissues and confirmed the presence of N^2 , 3- ϵ dG in hu-

man liver by mass spectrometric techniques. These findings suggested an endogenous pathway (fig. 1) for the formation of exocyclic adducts *via* lipid peroxidation products.

OXIDATIVE STRESS AND LIPID PEROXIDATION

Chronic inflammatory infection is one of the sources of free oxygen radicals and also leads to nitric oxide synthase (NOS) induction and therefore to NO synthesis. Oxidative stress processes enhance the generation of such reactive oxygen species as $O_2^{\cdot-}$, H_2O_2 and OH.

The most reactive molecule is the hydroxyl radical. Its production can be increased in response to accumulation of free Cu and Fe ions in tissues (mainly in the liver) which is known to occur in some procancerogenic diseases, Wilson disease and primary hemochromatosis.

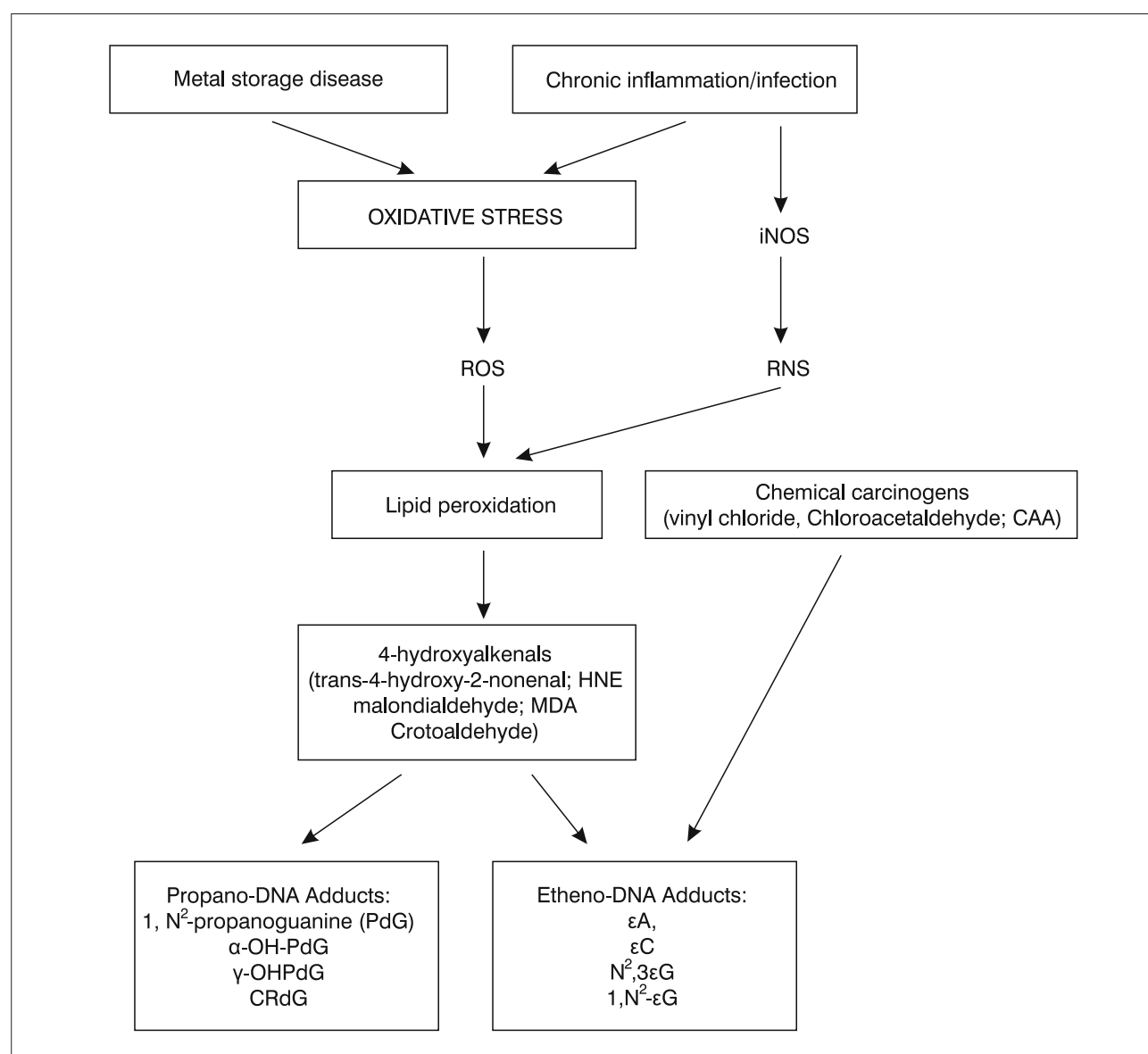
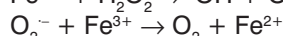
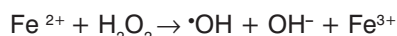
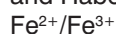


Fig.1. Proposed scheme of carcinogenic factor leading to oxidative stress-induced reactive oxygen species (ROS) and nitrogen (RNS) species; which cause exocyclic DNA-base damage. Where iNOS; inducible nitric oxide synthase (7).

These transient metal ions participate in Fenton and Haber-Weiss reactions to produce hydroxyl radicals:



and Haber Weiss reactions;



Poly-unsaturated lipids, components of lipids bilayers which surround various subcellular micro-environments, are one of the possible targets of free radical attack. The toxic effect of lipid peroxidation (LPO) is connected with the loss of cell membranes function and cell viability (9). Lipid peroxidation occurs in three steps; initiation, propagation and termination, and yields stable products which can either directly react with nucleic acids, or be further metabolized into more reactive compounds (tab. 1).

HNE – A MAJOR LPO PRODUCT OF DIVERGENT REACTIVITY

The major hydroxyalkenal is trans-4-hydroxy-2-nonenal (HNE). Its concentration in human plasma and tissues ranges between 0.1-3.0 μM and can increase to 10 μM in conditions of oxidative stress. HNE is genotoxic in bacterial and mammalian cells. At physiological concentrations, HNE increases the frequency of micronuclei, chromosomal aberrations, sister-chromatid exchanges and point mutations in mammalian cells. HNE is also a potent inducer of the SOS response in *Escherichia coli* at very low concentrations (0.1-1 μM) (11). Moreover, it exerts a clastogenic effect in human cells, possibly via inactivation of functional SH groups in DNA polymerases. HNE also participates in the regulation of many pathophysiological processes, such as inflammation, cell differentiation, apoptosis, liver fibrosis and carcinogenesis, by formation of adducts in reaction with cellular phospholipids, proteins and nucleic acids. Generation of HNE from low-density lipoprotein and its subsequent binding to apolipoprotein AI and AII,

apolipoprotein B and apolipoprotein E (12), has been implicated in the pathogenesis of atherosclerosis (13). Other various adverse biological effects of HNE include inhibition of RNA and DNA synthesis stimulation of neutrophil migration, enzyme inhibition, activation of stress-signaling pathways via transcription factors and kinase pathways, calcium homeostasis disturbances, inhibition of mitochondrial respiration, and morphological changes (14).

CHEMICAL STRUCTURE OF HNE

HNE is an extraordinarily reactive compound containing three functional groups: a conjugated system of C = C double bond, a C = O carbonyl group which provides a partial positive charge to carbon three and a OH group, the which inductive effect of which is increased at carbon four (fig. 2), (15).

Therefore, the nucleophilic attack, e.g., by thiol or amino groups occurs primarily at carbon three and secondarily at the carbonyl carbon one. Furthermore, HNE shows chirality at carbon four, which may also be biologically relevant. Crouzet et al. (16) have shown that (R)- and (S)-HNE are enantioselectively metabolised in rats, and the (S)-enantiomer shows preferential cytotoxicity in normal rat liver cytosol.

FORMATION OF EXOCYCLIC ADDUCTS TO DNA BASES VIA LIPID PEROXIDATION AND CHEMICAL CARCINOGENS

Cyclic etheno adducts were first found to be formed in DNA as a consequence of exposure to environmental carcinogens, such as vinyl chloride and its metabolites, chloroethylene oxide and chloroacetaldehyde, produced after oxidation by cytochrome P450 enzymes (CYP). In addition, anticancer drugs or mucochloric acid, generated from chlorine present in tap water, can also contribute to the formation of etheno bases in DNA. Another possible source of exocyclic adducts is X-ray. X-ray causes fragmentation of carbohydrates, one of the product being phosphoglycoaldehyde, shown to add to deoxyguanosine to form etheno adducts. In tissues of non-exposed rodents and humans several exocyclic adducts have been found and quantified. These include 1,N²-propanodeoxyguanine (PdG) and etheno DNA adducts, such as 1,N⁶-ethenoadenine (1,N⁶- ϵA), 3,N⁴-

Table 1. Main carbonyl products of lipid peroxidation separated and stimulated by carbon tetrachloride or ADP-iron in isolated rat hepatocytes and rat liver microsomal suspensions (10).

Polar carbonyls:
Malondialdehyde
Acrolein
Crotonaldehyde
Non-polar carbonyls:
Hydroxyalkenals:
4-hydroxy-2-hexenal
4-hydroxy-2-nonenal
4-hydroxy-2,5-dienal

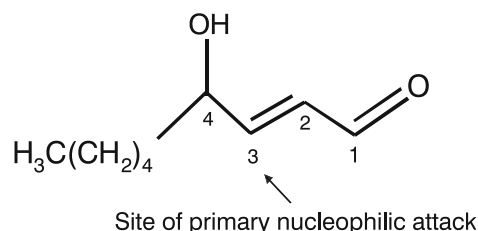


Fig. 2. The chemical structure of 4-hydroxy-2-trans-nonenal (HNE).

ethenocytosine ($3,N^4\text{-}\epsilon\text{C}$), $N^2,3\text{-}\epsilon\text{G}$ and $1,N^2\text{-}\epsilon\text{G}$). It has been postulated that these lesions are formed in mammalian tissues under conditions of lipid peroxidation. The precise mechanism of formation of these lesions is unknown, although formation of etheno adducts, was observed *in vitro* when deoxyguanosine was exposed to 2,3-epoxy-4-hydroxynonenal (EH). Hydroxynonenal was shown *in vitro* to bind to deoxyguanosine and form $1,N^2$ -propano-adduct with a hexyl side chain. These adducts were found in rodent and human DNA in the range of $1.8\text{--}15.8$ adducts/ 10^8 nucleotides (17).

Ethenoadenine and ethenocytosine have been detected, by immunoaffinity/ ^{32}P postlabeling, in DNA of untreated rodents and humans at levels ranging from 0.043 to 35 ϵA per 10^8 unmodified adenine residues and 0.05 to 24 ϵC per 10^8 unmodified cytosines. Using gas chromatography/mass spectrometry (GC/MS) technique Ham and co-workers estimated $N^2,3\text{-}\epsilon\text{dG}$ to range between $50\text{--}1720$ adducts/ 10^6 normal dGuo bases in calf thymus.

VINYL CHLORIDE AS AN INDUCER OF ETHENO – DNA ADDUCTS

Vinyl chloride -induced mutagenesis depends on oxidative metabolic activation. In mammalian cells, vinyl chloride is activated into chloroethylene oxide by cytochrome P450-dependent microsomal monooxygenases. Chloroethylene oxide binds directly to nitrogen atoms of DNA bases, forming adducts in the following quantitative order: $N^7\text{-(2-oxoethyl)guanine} \gg 1,N^6\text{-}\epsilon\text{A} > \text{hydroxyethanoguanine} > N^2,3\text{-}\epsilon\text{G} > 3,N^4\text{-}\epsilon\text{C} > 1,N^2\text{-}\epsilon\text{G}$. Hydroxyethanoguanine undergoes

further rearrangement to $N^2\text{-(2-oxoethyl)guanine}$. Alternatively, chloroethylene oxide rearranges to form chloroacetaldehyde (CAA). CAA binds to adenine and cytosine in DNA, forming mainly hydroxyethano derivatives, which subsequently dehydrate to $1,N^6\text{-}\epsilon\text{A}$ and $3,N^4\text{-}\epsilon\text{C}$ (fig. 3).

In poliribonucleotides, the conversion of hydroxyethanocytosine to ϵC occurs with a calculated half-life of 15 h, and conversion of hydroxyethanoadenine to ϵA occurs with a half-life of 1.4 h at 37°C at pH 7.25 (18). Formation of hydroxyethano guanosine is also possible, although these derivatives were not detected upon treatment of nucleic acids or nucleosides with CAA. These hydrated forms may also be formed in DNA of animals exposed to vinyl compounds, but their biological significance is poorly elucidated. Reaction of CAA with guanine in DNA favours the formation of $N^2,3\text{-}\epsilon\text{G}$; $1,N^2\text{-}\epsilon\text{G}$ is also formed but with at least 100 -fold lower efficiency (18). If N1 is not blocked by hydrogen bonding like in free nucleosides, the CAA induced formation of $1,N^2\text{-}\epsilon\text{G}$ prevails over that of $N^2,3\text{-}\epsilon\text{G}$. The quantitative relationship among etheno adducts induced by CAA in double-stranded DNA was reported to be the following: $3,N^4\text{-}\epsilon\text{C} \geq 1,N^6\text{-}\epsilon\text{A} > N^2,3\text{-}\epsilon\text{G} \gg 1,N^2\text{-}\epsilon\text{G}$ (18).

A more detailed analysis of the accumulation of ethenobases in relation to the length of exposure was done on rat liver, lung and kidney. ϵC was found to accumulate in the three organs, whereas ϵA accumulated in the liver but not in the kidney. In the lung, a steady-state level of ϵA was attained after 2 weeks of exposure. A more specific, high-resolution GC-MS technique also became available for measuring $N^2,3\text{-}\epsilon\text{G}$. Combining immunoaffinity purification with this technique, Swenberg (19)

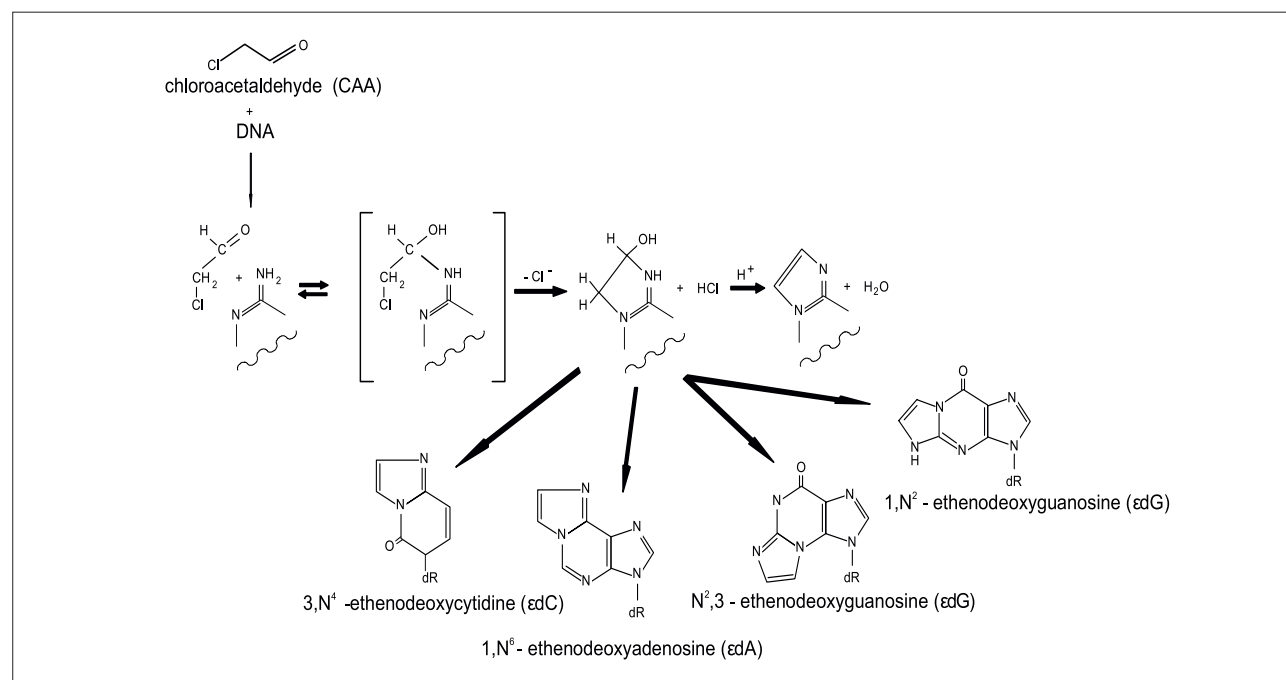


Fig. 3. Schematic presentation of the hydroxyethano derivatives formation induced by vinyl chloride.

measured $N^2,3\text{-}\epsilon\text{G}$ in liver DNA from adult rats exposed to different doses of vinyl chloride and mice exposed for 1 year to vinyl fluoride. These authors observed dose-related effects on the levels of $N^2,3\text{-}\epsilon\text{G}$ formed in hepatic DNA (19) and also measured background levels of $N^2,3\text{-}\epsilon\text{G}$ ranging from 6×10^{-8} to 7×10^{-7} (molar ratio $N^2,3\text{-}\epsilon\text{G}$ / G in a series of 12 DNA samples from human liver). They detected similar levels in unexposed rats. All this places etheno DNA adducts as one of the most important components of the carcinogen/oxidative stress pathway leading to genome instability, cancer and other degenerative processes.

MUTATIONS INDUCED BY ETHENO AND PROPANO-DNA ADDUCTS

In bacteria and mammals (simian kidney cells), etheno and propano adducts induce base substitutions, frameshift mutations as well as sister chromatid exchanges and chromosomal aberrations. Etheno DNA adducts are moderate inhibitors of DNA synthesis, both *in vitro* and in bacterial and mammalian cells. However replicative DNA polymerases tend to incorporate non-cognate nucleotides opposite etheno-adducts, which leads to mutations with a frequency, dependent on the source of DNA polymerase. Mutagenic properties of propane adducts have been established for malondialdehyde, acrolein, crotonaldehyde and trans -4-hydroxy-2-nonenal (HNE) (tab. 2).

DNA REPAIR

To counteract deleterious consequences of DNA damage, the cells developed several repair mechanisms which eliminate from genomes mis-instructive or non-instructive elements, as well as seal DNA breaks. Repair of exocyclic, (ethano and propano DNA adducts is realized by different systems of DNA repair, mainly by base excision repair pathway (Gros et al., 2003), initiated by DNA glycosylases which cleave out the damaged base and initiate the synthesis step and by nucleotide excision repair in mammalian cells in which a larger fragment of damaged DNA strand is removed (12-13 nucleotides in *E.coli*, 24-32 nucleotides in eucaryota).

Abbreviations: $\alpha\text{-OH-PdG}$: 6R and 6S isomers of 3H-6-hydroxy-3-(β -D-2'-deoxyribofuranosyl)-5, 6, 7, 8-tetrahydropyrido[3, 2-a]purine-9-one, β **compound**: 4-amino-5-(imidazol-2-yl)imidazole, $\gamma\text{-OH-PdG}$: 8R and 8S isomers of 3H-8-hydroxy-3-(β -D-2'-deoxyribofuranosyl)-5,6,7,8-tetrahydropyrido[3,2-a]purine-9-one, **CV**: vinyl chloride, ϵ -etheno, ϵA : 1, N^6 -etheno-adenine, ϵC : 3, N^4 -ethenocytosine, ϵG : 1, N^2 -ethenoguanine, $N^2,3\text{-}\epsilon\text{G}$: $N^2,3$ -ethenoguanine, **iNOS**: inducible nitric oxide synthase, **MDA**: malondialdehyde, **M1-dA**: N^6 -(3-oxo-propenyl) deoxyadenosine, **M1-dC**: N^4 -(3-oxopropenyl) deoxycytidine, **M1-dG**: (pyrimido[1,2 α]purin-10(3H)-one), **NOS**: nitric oxide synthase, **PUFAs**: polyunsaturated fatty acids, **RNS**: reactive nitrogen species. □

Table 2. The types of base changes induced by ethenobases and propanobases observed *in vitro* in *E.coli* and mammalian cells.

Lesion	Base changes		
	<i>In vitro</i>	<i>E.coli</i>	Mammalian cells
ϵA	A→G, A→T > A→C	A→G > A→C, A→T	A→G > A→T, A→C
β	A→T > A→C	A→G, A→C, A→T	Not Determined
ϵC	C→A, C→T > C→G	C→T, C→A	C→A, C→T > C→G
$\epsilon\text{C}\cdot\text{H}_2\text{O}$	No incorporation	C→T	Not Determined
$N^2,3\epsilon\text{G}$	G→A	G→A	G→T, G→A
1, $N^2\text{-}\epsilon\text{G}$	G→T, G→C	G→T, G→C, G→A	G→A > G→T
HO-ethanoG	G→T, G→C	G→T, G→C, G→A	Not Determined
Malondialdehyde M_1G	G→T, C→T > A→G $\text{M}_1\text{G}\rightarrow\text{A}$, $\text{M}_1\text{G}\rightarrow\text{T}$, $\text{M}_1\text{G}\rightarrow\text{C}$		
acrolein $\alpha\text{-OH-PdG}$ $\gamma\text{-OH-PdG}$	G→T, C→T > A→G		
crotonaldehyde	C→A, G→T, G→C, C→G, G→A, C→T > A→T > > T→A		
Trans-4-hydroxy-2-nonenal (HNE)	G→T, C→A		

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Received: 27.07.2012

Accepted: 16.08.2012

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